## SYNTHESIS OF 4-(1-METHYL-2-PYRROLIDINYL)ISOQUINOLINE UNDER PHYSIOLOGICAL CONDITIONS : A MODEL FOR THE BIOSYNTHESIS OF 4-PYRROLIDINYLISOQUINOLINE ALKALOIDS

## Edward Leete

## Natural Products Laboratory,<sup>1</sup> School of Chemistry, University of Minnesota, Minneapolis, Minnesota, 55455 USA

Abstract.-1,2,3,4-Tetrahydroisoquinoline-3-carboxylic acid was oxidatively decarboxylated with sodium hypochlorite, affording a dihydroisoquinoline which on condensation with N-methyl- $\Delta^{1-}$  pyrrolinium acetate, and subsequent aerial oxidation yielded 4-(1-methyl-2-pyrrolidinyl)iso-quinoline.

Arenine (6, R = H) and macrostomine (6, R = Me) are novel isoquinoline alkaloids which have been isolated from *Papaver arenarium*<sup>2</sup> and *P. macrostomum*<sup>3</sup> respectively. Although no biosynthetic experiments have been carried out on these alkaloids, which contain a pyrrolidine ring at the 4-position of the isoquinoline nucleus, we consider that these alkaloids are formed from norlaudanosoline-3-carboxylic acid (1) which is formed from dopa and 3,4-dihydroxyphenylpyruvic acid. It is suggested that this compound undergoes an oxidative decarboxylation af-



fording the 1,4-dihydroisoquinoline (2) which tautomerizes to the 1,2-dihydroisoquinoline (4). This enamine then condenses with the N-methyl- $\Delta^1$ -pyrrolinium salt (3), which is presumably formed from ornithine. This reaction is analogous to one involved in the biosynthesis of nicotine where (3) condenses with a dihydropyridine.<sup>4</sup> Oxidation and suitable methylations of (7) then yield the alkaloids (6). We have been able to simulate these reactions in this proposed scheme, starting with 1,2,3,4-tetrahydroisoquinoline-3-carboxylic acid (8).5 This compound was dissolved in dilute NaOH (pH 10) and treated with an equivalent of NaOC1. After 5 minutes an equivalent of  $(3)^6$  was added and the solution (pH 7.5) stirred for 7 days in an open beaker at room temperature. A CHCl3 extract of this solution yielded a mixture of isoquinoline and 4-(1-methyl-2-pyrrolidinyl) isoquinoline (9). This latter compound which could also be named benzo[d]nicotine was obtained as a pale yellow oil, IR (neat) 2830 cm<sup>-1</sup> (NMe); UV (95% EtOH)  $\lambda_{max}$  (Log  $\epsilon$ ) 219 (4.68), 262 (3.36), 272 (3.40), 285 (3.31), 310 (3.38), 318 sh (3.40), 323 (3.52) MS m/e 212 (M<sup>+</sup>), 84 (N-methylpyrrolinium ion); <sup>13</sup>C NMR (CDCl<sub>3</sub>) ppm from Me<sub>4</sub>Si 151.6 (C-1), 141.1 (3), 134.7 (4a), 131.8 (6), 129.8 (8a), 128.3 (8), 128.1 (7), 126.6 (5), 123.3 (4), 66.9 (2'), 57.2 (5'), 40.8 (NMe), 34.0 (3'), 22.9 (4').<sup>7</sup> This product was identical with a sample prepared from ethyl isoquinoline-4-carboxylate<sup>8</sup> using a method analogous to Späth's synthesis of nicotine,<sup>9</sup> and the synthesis of benzo[e]nicotine from ethyl quinoline-3-carboxylate.<sup>10,11</sup> Brevicolline (5), using similar physiological conditions, has been recently prepared from 1-methyl-1,2,3,4-tetrahydro-B-carboline-3-carboxylic acid and (3).12

This investigation was supported by a research grant GM-13246 from the National Institutes of Health.



## References

- 1. Contribution No. 171 from the laboratory.
- I. A. Israilov, M. A. Manushakyan, M. S. Yunusov, V. A. Mnatsakanyan, and S. Ya Yunusov, 2.
- Khim. Prir Soedin., 14, 417 (1978). V. A. Mnatsakanyan, V. Preininger, V. Simanek, J. Jurina, A. Hlasek, L. Dolejs and 3. F. Santavy, Collect. Czech. Chem. Commun., 42, 1421 (1977).
- 4.
- E. Leete, <u>Bloorg. Chem., 6</u>, 273 (1977). P. L. Julian, W. J. Karpel, A. Magnani, and E. W. Meyer, <u>J. Am. Chem. Soc</u>., 70, 180 (1948). 5.
- 6.
- E. Leete, J. Am. Chem. Soc., 89, 7081 (1967). Assignments were made by comparison with the <sup>13</sup>C NMR spectra of nicotine<sup>4</sup> and isoquinoline (Spectrum No. 334, J. F. Johnson and W. C. Jankowski, Carbon-13 NMR Spectra, John Wiley and Sons, Inc., New York, NY (1972).

- 8. F. T. Tyson, J. Am. Chem. Soc., 61, 183 (1939).
  9. E. Späth and H. Bretschneider, Chem. Ber., 61, 327 (1928).
  10. B. K. Nandi, J. Indian Chem. Soc., 17, 285 (1940).
  11. A. Ide, K. Matsumori, and H. Watanabe, Nippon Kagaku Zasshi, 91, 578 (1970).
- 12. E. Leete, J.C.S., Chem. Commun., in press (1979).

(Received in USA 17 August 1979)